Simple Averaging Procedure for Aromatic Ring System: Mean Molecular Fragment of Tryptophan Metabolites

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Abstract

A simple averaging procedure of bond lengths and angles applicable to an aromatic ring system was applied to the crystal structures of the tryptophan metabolites analyzed so far, as an example. Statistical analysis showed no marked effect of the 5-substituent on the molecular dimensions of the 3-substituted indole ring in tryptophan metabolites. The standard bond lengths and angles of the 3-substituted indole ring were calculated according to the statistical averaging procedure outlined here and using data from 35 tryptophan metabolites; the interatomic correlations of the indole ring have been discussed based on these values. The result suggests the validity of the D^2 statistic for averaging the planar ring structure in aromatic systems.

1. Introduction

The exact determination of molecular geometry is of vital importance to our understanding of chemical structure and bonding behavior. It also gives clues to the electronic nature of a molecule and its intrinsic propensity, by which it can play a unique role. The 'standard' bonding parameters, on the other hand, enable evaluation of the effect of a substituent or the accuracy or the significance of differences in experimental data.

Much experimental data have been derived from X-ray and neutron diffraction measurements, and structural parameters of over 140000 organocarbons have been accumulated in the Cambridge Crystal-lographic Data Centre (Allen *et al.*, 1991). Methods have been developed to estimate the average molecular dimensions (AMD) for aliphatic organic compounds and the validity of these methods has been tested (Chakrabarti & Dunitz, 1982; Schweizer & Dunitz, 1982; Taylor & Kennard, 1983, 1985, 1986; Allen *et al.*, 1987). There appear to be rather few studies (Taylor & Kennard, 1982) concerning the estimation of AMD for aromatic compounds in which all bonding parameters cannot be treated as independent.

As far as the aromatic ring structures are concerned, there would be considerably strong correlation among bond lengths and angles because of their strict restriction by the ring formation; each of them is closely related with one another in a planar ring. It is supposed that these interrelations are rather inversely proportionate to each other in the closed-ring structure. In statistical analyses for such ring structures, therefore, the parameters (bond lengths/angles) in planar rings should be treated as multivariates. The procedure presented here could detect the ring deformations as multivariate outliers and thus could quantitatively describe the whole deviation of each structure from the mean. This is the main advantage of averaging the ring structures and is superior to other trials so far.

In order to investigate a possible relationship between the metabolic pathway of tryptophan and the conformation of each metabolite, on the other hand, we have been analyzing crystal structures of a series of tryptophan metabolites. A set of standard bond lengths and angles for the tryptophan indole ring is very useful to estimate the accuracy of the analyzed bonding parameters and to consider the accurate electronic/biological propensity of each tryptophan metabolite. However, no systematic and satisfactory AMD estimation method has been reported so far. Under such circumstances we developed an averaging procedure for the indole aromatic ring. In order to estimate the AMD of tryptophan metabolites, the bond lengths and angles of the planar ring structures were summed up en bloc and parameters of multivariate outliers were excluded from the estimation.

2. Data set

An atomic numbering of the common moiety of tryptophan metabolites is shown in Fig. 1. A bibliographic list of all structures used in the analysis is given in Table 1, where all trytophan metabolites reported so far, except for the metal complexes and the 4-substituted nonmetabolites, are

 R^{1}/R^{2} substituents

Table 1. Data set of 46 tryptophan metabolites with Table 2. Distribution of data and statistical values (n = 46)

					Reference		0	а.	<i>a</i> 2	a	а.	Mean	σ	OD
No.	R^1	R^2	Form	CSD code	(coden)	NI C2	1 280	1 260	1 275	1 205	1 462	1 276	0.022	0.000
			F		DDA 642 122	NI-C2	1.280	1.309	1.375	1.385	1.403	1.370	0.023	0.008
1		$-CH_2NH_2$	riee C-h	TOVETAIO	DDA 343 123	$NI = C_{3}$	1.323	1.370	1.373	1.360	1.452	1.3/0	0.010	0.003
2		$-CH_2NH_3$	Salt	TRIPIALO	BCSJA8 40 2481	$C_2 = C_3$	1.309	1.337	1.304	1.3/1	1.419	1.304	0.010	0.007
3		$-CH_2NH_3$	Complex	TRIFIC	ACDCAR 30 1041	$C_3 = C_9$	1.324	1.420	1.433	1.444	1.313	1.455	0.024	0.008
4			Complex	INTER	ICCCAT 259	$C_4 - C_9$	1.370	1.397	1.405	1.412	1.450	1 377	0.014	0.0073
5			Complex	TDATAA	ACRCAR 25 1642	$C_4 - C_5$	1.331	1.307	1.370	1.303	1.450	1 200	0.019	0.000
7		$-CH NH^+$	Complex	TPADAA	ACDCAR 33 1042	$C_{5} = C_{0}$	1.332	1.367	1 275	1 3 8 3	1.401	1.374	0.023	0.0103
8		$-CH NH^+$	Complex	GANEOO	ACAPCT 41 530	C_{7}	1.250	1 386	1 303	1.303	1 474	1 395	0.019	0.007
ğ		-CH-NH ⁺	Complex	DOTXUEIO	IACSAT 110 2286	C8-C9	1 370	1 399	1 406	1 412	1 457	1 405	0.015	0.0065
10		-CH(COOH)NH ⁺	Salt	TRYPTC	BCSIA8 39 2369	00 07	1.570	1.577	1.400		1.107	1.105	0.012	0.0000
11		$-CH(COOH)NH^+$	Salt	TRYPTDIO	CUSCAM 36 139	$C_{2} = N_{1} = C_{8}$	105.0	108.1	108.6	109.2	110.6	108.6	11	0.55
12		-CH(COOH)NH ⁺	Salt	TRYPTF10	ACSAA4 27 471	N1 - C2 - C3	108.7	109.7	110.3	110.7	114.9	110.3	1.1	0.5
13		-CH(COOH)NH ⁺	Salt	TPTPCM	ACBCAR 30 1841	$C_2 - C_3 - C_9$	102.4	105.6	106.2	106.6	107.4	106.1	0.9	0.5
14		-CH(COOH)NH ⁺	Salt	ТРҮРТВ	ACBOCV 34 559	C3-C9-C8	102.3	106.9	107.3	107.7	110.9	107.4	1.3	0.4
15		-CH(COOH)NH ⁺	Free		CPBTAL 41 433	C4-C9-C8	116.4	118.7	119.4	120.0	124.5	119.5	1.5	0.65
16		$-CH(COO^{-})NH_{1}^{+}$	Free	OOOBTP01	ACBOCV 34 559	C5-C4-C9	113.8	117.3	118.5	119.0	120.2	118.1	1.5	0.85
17		$-CH(COO^{-})NH_{1}^{+}$	Free		CPBTAL 41 433	C4C5C6	119.2	120.9	121.4	122.2	125.4	121.7	1.4	0.65
18		$-CH_2N(CH_3)_2$	Free	DMTRYP	ACBCAR 28 3075	C5-C6-C7	116.4	120.6	121.4	122.3	124.2	121.3	1.4	0.85
19		$-CH_2N(CH_3)_2$	Free	DMTPYP	ACBCAR 28 3075	C6-C7-C8	112.8	116.9	117.4	118.1	122.7	117.4	1.5	0.6
20		-соон	Free	INACET	ACCRA9 17 496	C7-C8-C9	117.6	121.3	122.0	122.8	127.7	122.0	1.4	0.75
21		-COOH	Complex	IACNCA	BCSJA8 51 1118	N1-C8-C9	102.9	106.9	107.5	108.0	111.8	107.5	1.5	0.55
22		-COOH	Free	INACET01	ACBCAR 38 2534	N1-C8-C7	125.8	129.9	130.6	131.1	135.5	130.5	1.4	0.6
23		-C00-	Complex	IAAMTA	ACBCAR 32 3235	C3-C9-C4	128.7	132.7	133.3	133.8	135.5	133.1	1.2	0.55
24		COO	Complex		ABBIA4 200 492									
25		-COO-	Complex	BAKMOP	ACBCAR 37 2117									
26		-COO-	Complex	MEADIN10	BICHAW 22 3571	structures	were	use	d fo	or th	ie pi	resent	t sta	tistica
27		-C00-	Complex	CEBZUE10	JACSAT 110 2286	treatment.					_			
28		-C00-	Complex	CEBZUE10	JACSAT 110 2286	A fton con		f the	16		maa fa		h man	mata
29		-C00-	Complex	GATSAV	JACSAT 110 2286	Aller sor	ung c	or the	40 SL	ructu	res ic	or eac	n para	amete
30		-C00-	Complex	GATSAV	JACSAT 110 2286	(bond lengt	h and	angle	e), the	e quai	rtiles,	q_0, q	$_{1}, q_{2},$	q_3 and
31	-OH	$-CH_2NH_3^+$	Complex	HTRCRS	ACCRA9 19 713	a, were of	otaine	d wh	ich c	orres	nond	to the	e min	imum
32	-OH	-CH ₂ NH [*]	Salt	SERPIC	ACBCAR 28 82	25 50 on	d 75	noro	antila	0 07	d m	vim		f oool
33	-OH	-CH ₂ NH ₃	Salt	SERHOX	ACAPCT 32 267	25, 50 and	u 75	perc	entre	s, ai	iu ma		in o	i taci
34	-OH	$-CH(COO^{-})NH_{3}^{+}$	Free	HIRYPIIO	BCSJA8 46 2475	parameter,	resp	ective	ely.	A li	ist o	f the	ese v	values
35	-OH	$-CH_2N(CH_3)_2$	Free	BUFTEN	ACBCAR 28 3219	together w	ith th	e me	an. s	standa	ard d	eviati	on (τ) and
27	-0n	$-CH_2N(CH_3)_2$	Free	BUFIEN	ACBCAR 28 3219	quartila da	viatio		<u> </u>	(0	a)/) (D	ad 1	08611
37			Free	JEIDUK	ACSCEE 40 2420	qualitie dev			ש) =	$(q_3 -$	$- \frac{q_1}{2}$	2 (10	au, i	.900)]
30	-0CH	$-CH_2NH_2$	Complex	MIAIRIP	ACBCAR 30 33	is given in	Table	e 2. T	he sr	nall (ζD va	alues	of the	e bon
40		$-CH_2NH^+$	Complex	ΙΔΔΜΤΔ	ACBCAR 32 3233	lengths (0.0	005-0	.010	Ă) an	id an	gles (0.4-0).8°)	for al
41		-CH-NHCOCH	Free	MELATN	CMI TAG 1139	the 16 st	ructu.		adicat	to th	at th		fact	of 5
42	-0CH	-CH_NHCOCH	Free	MELATNOI	ACBCAR 30 99			CS 11	iuicai	ι υ	מנ 11			
43	OCH-	-CH_NHCOCH.	Free	MELATN02	ACBOCV 28 564	substitution	on th	ne mo	lecul	ar dii	mensi	ons o	t the	indol
44	-OCH	$-CH_2NH(CH_2)^+$	Salt	MOTYPT	ACBCAR 27 411	ring is not	as sig	nifica	nt as	has l	been a	suppo	sed.	On th
						0	- 6							

Salt: HCl, HBr, formate, oxalate or picrate.

-OCH₃ -COOH

-OCH, -COO-

45

46

included.* All tryptophan metabolites used in this paper are 3-substituted (numbers 1-30) and 3,5disubstituted (numbers 31-46) derivatives. Thus, 46

Free

MXINAC

Complex MIAMTA

BCSJA8 48 536

ACBCAR 32 3235

* A list of bond lengths and bond angles of 46 structures of tryptophan metabolites has been deposited with the IUCr (Reference: OA0001). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.



Fig. 1. Chemical structure and atomic numbering of the common moiety of trytophan metabolites.

eter and JM. ach les, and 6)], ond all 5lole ring is not as significant as has been supposed. On the other hand, some bond lengths and angles show threefold larger σ values than the corresponding QD values, indicating that some data unsuitable for the estimation of AMD are included in the present data set. For statistical treatment of reliable data, therefore, it is of utmost importance to develop an appropriate method for the detection and exclusion of such multivariate outliers. This is one of the main objectives of the present study.

3. Methods

In order to obtain the mean ring structure of tryptophan metabolites, we used the D^2 statistic as a measure of the distance between the two populations (Mahalanobis, 1936; Pillai, 1985), and this statistic is also applicable as a measure of the distance between each sample structure and its mean in the present study.

When **X** and μ denote the observed and mean vectors of p geometrical variables, respectively, the distance

	5-sub	5-substituted $(n - 16)$			Unsubstituted $(n-30)$				
	Range	Mean	σ	Mean	σ	1-10	P		
N1-C2	1.352-1.409	1.376	0.015	1.375	0.027	0.163	0.87		
N1-C8	1.362-1.394	1.375	0.008	1.376	0.019	0.247	0.81		
C2-C3	1.346-1.389	1.366	0.009	1.363	0.018	0.731	0.47		
C3-C9	1.403-1.469	1.438	0.015	1.434	0.028	0.635	0.53		
C4-C9	1.370-1.418	1.400	0.013	1.407	0.015	1.593	0.12		
C4-C5	1.351-1.416	1.375	0.015	1.377	0.021	0.336	0.74		
C5-C6	1.379-1.419	1.405	0.010	1.395	0.027	1.790	0.08		
C6C7	1.352-1.398	1.373	0.013	1.375	0.022	0.394	0.70		
C7–C8	1.363-1.431	1.394	0.018	1.396	0.021	0.326	0.75		
C8—C9	1.379-1.415	1.403	0.009	1.407	0.018	1.020	0.31		
$C_{2}-N_{1}-C_{8}$	105.6-110.6	108.6	1.1	108.6	1.0	0.0	1.0		
N1 - C2 - C3	108.9-113.4	110.3	1.1	110.4	1.2	0.284	0.78		
C2-C3-C9	104.0-107.1	106.1	0.8	106.1	1.0	0.0	1.0		
C3-C9-C8	105.3-110.4	107.2	1.1	107.5	1.4	0.746	0.46		
C4-C9-C8	116.4-122.9	119.9	1.4	119.2	1.5	1.540	0.13		
C5-C4-C9	116.1-119.7	118.0	1.0	118.1	1.7	0.247	0.81		
C4-C5-C6	119.6-124.1	121.7	1.2	121.6	1.6	0.221	0.83		
C5-C6-C7	118.9-123.1	120.9	1.2	121.5	1.4	1.417	0.16		
C6-C7-C8	116.2-118.8	117.8	0.7	117.2	1.7	1.659	0.10		
C7-C8-C9	120.6-123.9	121.7	0.9	122.2	1.7	1.353	0.18		
N1-C8-C9	104.2-111.0	107.8	1.5	107.4	1.4	0.888	0.38		
N1-C8-C7	128.2-131.9	130.5	1.1	130.4	1.6	0.224	0.82		
C3-C9-C4	131.8-133.7	132.9	0.5	133.3	1.4	1.379	0.18		

 (D_i^2) of the *i*th sample from the mean in *p*-dimensional space is defined as

$$D_i^2 = (\mathbf{X}_i - \boldsymbol{\mu})' \boldsymbol{\Sigma}^{-1} (\mathbf{X}_i - \boldsymbol{\mu}), \qquad (1)$$

where $(\mathbf{X}_i - \boldsymbol{\mu})'$ represents the transposed matrix of difference vector $(\mathbf{X}_i - \boldsymbol{\mu})$ and $\boldsymbol{\Sigma}^{-1}$ is the inverse matrix of the $p \times p$ covariance matrix $\boldsymbol{\Sigma}$. Each component of $\boldsymbol{\Sigma}$, V_{jk} (the covariance of the *j*th and *k*th variables), is defined as

$$V_{jk} = [1/(n-1)]\Sigma(X_j - \mu_j)(X_k - \mu_k), \qquad (2)$$

where n is the number of samples.

The D^2 statistic (D_i^2 in 1) is well known as Mahalanobis' squared distance (Mahalanobis, 1936; Pillai, 1985) and obeys the χ^2 distribution for *p* degrees of freedom. In this study we attempted to determine multivariate outliers and to quantitatively estimate the deviation from the mean ring structure by applying this D^2 statistic to the indole rings of tryptophan metabolites. A conventional statistical test, the *t*-test, was also used for the comparison of the bonding parameters between two groups, *i.e.* 5-substituted and unsubstituted structures. For *t*, χ^2 and *r* values for the degree of freedom ν at the level of significance α , the expressions of $t(\nu, \alpha)$, $\chi^2(\nu, \alpha)$ and $r(\nu, \alpha)$ were used.

4. Results and discussion

4.1. Effect of 5-substitution

In order to assess the effect of 5-substitution on the molecular dimensions of tryptophan metabolites, all

bond lengths and angles were classified into two groups (numbers 1-30 and 31-46 in Table 1), and were separately treated for statistical calculations. A comparison of the results from both groups is given in Table 3. From Tables 2 and 3 it is found that all the structural parameters of 5-substituted derivatives are in the range of unsubstituted structures. Moreover, neither t(44,0.05) value >2.0154 nor p value <0.05 is found in Table 3, indicating that the molecular dimensions of indole rings of both groups are essentially in the same range. This is also clear from histograms of bond angles involving C5 (Fig. 2), where the bond angles of C5substituted compounds are distributed without specific propensity on the histogram of unsubstituted compounds; bond lengths of the C5-substituted structures also show almost the same distribution as those of the unsubstituted compounds. Thus, it could be concluded that the substitution at C5 does not affect the molecular dimensions of 3-substituted tryptophan metabolites at



Fig. 2. Distribution of bond angles (C5-C4-C9, C4-C5-C6) and C5-C6-C7. The filled, hatched and dotted bars represent unsubstituted (n = 30), 5-hydroxy- (n = 7) and 5-methoxy- (n = 9) substituted tryptophan metabolites, respectively.

all, so far as the C5-substitution is the hydroxyl or methoxyl group in the tryptophan metabolite. For the subsequent statistical treatment, therefore, data were treated *en bloc*.

4.2. D^2 statistic and multivariate outliers

Using equation (1), the D^2 statistic (Mahalanobis' squared distance) was applied to the bond lengths and angles of 46 structures in Table 1. The results are given in Fig. 3(*a*), where ten bond lengths constituting the indole ring were treated as variables. The results for 13 bond angles are given in Fig. 3(*b*). A slight difference in distribution could be observed between the bond lengths and angles, reflecting a certain degree of independence between them. Concerning the bond lengths, there are nine structures with $D^2 \ge 18.307 [= \chi^2(10, 0.05)]$ and



Fig. 3. Histograms of D^2 statistics using (a) 10 bond lengths and (b) 13 bond angles, where the ordinate and abscissa correspond to the number of observations and $D^2(\text{length})$ or $D^2(\text{angle})$, respectively, and (c) a scattering diagram of $D^2(\text{length})$ and $D^2(\text{angle})$. For the calculations, data from all 46 structures were used.

Fable	4.	Mea	n mol	ecular	fragment	of	tryptopl	han
met	abo	lites	(n = 3.5)	5) and	comparison	wit	h indole	

	Range	Mean	σ	Indol
NIC2	1 352-1 409	1 3757	0.0121	1 383
N1 - C2 N1 - C8	1 362 1 304	1 3738	0.0121	1.36.
$C_{1}^{-C_{3}}$	1.302-1.334	1.3738	0.0073	1 269
$C_{2} = C_{3}$	1.042-1.070	1 /3/8	0.0082	1.300
C_{1}^{-}	1.405-1.471	1.4027	0.0110	1.41
$C_4 - C_5$	1.370-1.434	1.4027	0.0119	1.302
$C_{4} = C_{5}$	1.331-1.409	1.3735	0.0128	1.374
$C_{5} = C_{0}$	1.374-1.419	1.39/0	0.0122	1.393
C_{1}^{-}	1.332-1.409	1 2020	0.0119	1.374
$C^{\prime} = C^{0}$	1.303-1.423	1.3930	0.0124	1.34.
Co-C9	1.362-1.420	1.4049	0.0091	1.40.
C2-N1-C8	107 3-110 6	108 77	0 74	1104
N1 - C2 - C3	108.7-111.5	110 19	0.64	106.4
$C_2 - C_3 - C_9$	105.0-107.4	106.27	0.63	109.9
C3-C9-C8	106.1-110.4	107.33	0.78	105.8
C4-C9-C8	116.4-121.1	119.25	0.93	118.
C5-C4-C9	114.6-120.2	118.44	1.11	120.
C4-C5-C6	119.6-124.8	121.50	1.16	117.3
C5-C6-C7	118.9-123.1	121.23	1.06	123.9
C6-C7-C8	115.1-118.8	117.51	0.81	116.2
C7—C8—C9	120.6-123.9	122.06	0.82	123.0
N1-C8-C9	104.2-108.5	107.40	0.84	107.1
N1-C8-C7	128.2-132.0	130.53	0.93	128.8
C3-C9-C4	131.1-135.5	133.42	0.80	136.
C3-C10	1.462-1.530	1.4999	0.0121	
C2-C3-C10	124.9-131.9	127.44	1.61	
C9-C3-C10	121.1-129.5	126.25	1.81	

seven structures with $D^2 \ge 23.209 \ [= \chi^2(10, 0.01)],$ whereas in the case of bond angles, there are 11 structures with $D^2 \ge 22.362 \ [= \chi^2(13, 0.05)]$ and seven structures with $D^2 \ge 27.688$ [= $\chi^2(13, 0.01)$]. In order to estimate the correlation between bond lengths and angles, a scattering diagram is shown in Fig. 3(c). where the linear correlation coefficient r for 44 degrees of freedom is 0.874. From these figures, we find that: (i) seven structures, which are judged as outliers at $\alpha = 0.01$, are the same after both calculations of bond lengths and bond angles; (ii) nine structures in D^2 (length), which should be excluded at $\alpha = 0.05$, are all included in 11 structures of D^2 (angle); (iii) although r = 0.874 indicates a relatively high correlation between $D^2(\text{length})$ and $D^2(\text{angle})$, its collinearity is clear in the outlier region (right-upper region of Fig. 3c), especially, and the normal region (left-lower region) shows a rather broad distribution (poor correlation) from the collinearity, reflecting also a certain degree of independence between bond lengths and angles. On the basis of the analyses of the D^2 statistic, 11 structures (numbers 2, 5, 11, 15-17, 26, 29-31 and 46 in Table 1), which showed the outlier of D^2 (angle) at $\alpha = 0.05$, were excluded from the data set in the subsequent statistical treatment.

4.3. Mean molecular fragment

Using data from the 35 structures, the bonding parameters for the mean molecular fragment (MMF) were calculated and the results are given in Table 4. The

sums of three bond angles around C3, C8 or C9 are all 360.0° and the sums of interior angles of indole fiveand six-membered rings are 540.0 and 720.0°, respectively.

In order to estimate the effect of 3-substitution on the bonding parameter of the indole ring, the averaged values were compared with the bond lengths and angles of the indole ring itself. Although there are four reports on the analysis of the indole crystal in the Cambridge Structural Database (CSD), only one analysis [indole: 9-ethyladenine complex (Kaneda & Tanaka, 1976)] is available concerning its bonding parameters (Table 4). Thus, statistical comparison with the AMD of 3substituted tryptophan metabolites is impossible. However, a general tendency concerning the effect of 3substitution on the indole ring can be discussed as follows. The bond lengths C3-C9, C4-C9 and C7-C8 appear to increase with the substitution at C3. Furthermore, the increase in the N1-C2-C3bond angle is concomitant with the decrease in C2-C3-C9 and C3-C9-C4 bond angles. A similar change could also be observed in the C4-C5-C6, C5-C6-C7 and C5-C4-C9 angles.

4.4. Evaluation of D^2 statistic

Since values obtained from the D^2 statistic reflect the cumulative deviation of bond lengths or angles $(\mathbf{X}_i - \boldsymbol{\mu})$ of each observation (\mathbf{X}_i) from the AMD $(\boldsymbol{\mu})$, these values could be compared with δ_{length} or δ_{angle} $(= \Sigma | X_j - \boldsymbol{\mu}_j |)$, in order to further evaluate the validity of the D^2 statistic. The scattering diagram of δ_{length} with respect to D^2 (length) for all 46 structures is shown in Fig. 4; a similar tendency is also obtained for $\delta_{\text{angle}} - D^2$ (angle) plots (data not shown). The correla-



Fig. 4. Scattering diagram of δ_{length} and $D^2(\text{length})$ using data from 46 structures. Distributions of δ_{length} 's and $D^2(\text{length})$ range from 0.031 to 0.411 and from 0.49 to 33.10, respectively.

tion coefficients r for $\delta_{\text{length}} - D^2(\text{length})$ and $\delta_{\text{angle}} - D^2(\text{angle})$ pairs were 0.925 and 0.840, respectively. These high correlations suggest clearly that the ring deformation and total deviation from the mean ring can be treated by the D^2 statistic, quantitatively. Based on this background, we used effectively the D^2 statistic to detect outliers and exclude them from further averaging procedure.

On the other hand, Taylor and coworkers (Taylor & Kennard, 1985; Allen *et al.*, 1987) have recommended the prescreening method on the basis of precision (e.s.d. for C—C bonds) of X-ray analysis, in which the data with e.s.d.'s > 0.010 Å are eliminated beforehand as unreliable low-precision data. In order to evaluate the relationship between the D^2 outliers and low-precision X-ray analyses, 46 structures were surveyed again and classified into four groups by two features, *i.e.* D^2 statistic (outliers or not) and X-ray precision (less or greater than 0.010 Å). The classification result is given in the following fourfold table as below

31	1	32	$\chi^2 = 21.360 [\gg \chi^2(1, 0.001) = 10.828].$
4	10	14	
35	11	46	

The high χ^2 value (= 21.360) obtained from the fourfold table test indicates that a considerably strong interrelation exists between two features. In fact, 10 of 11 D^2 outliers are found later to be from low-precision analyses (e.s.d. > 0.010 Å) by comparison with the original literature or AS-flag (Allen *et al.*, 1987; Allen *et al.*, 1991) in the CSD database. On the contrary, sample number 26 (MEADIN10 in CSD code, AS-flag=2) is only one D^2 outlier from high-precision X-ray analyses. As shown in the above fourfold table, there are 14 structures from low-precision analyses, in which 10 structures are discriminated as outliers only by the calculation of D^2 values. In other words, the D^2 statistic could detect 10 out of 14 inaccurate structures, based on the total deviation from the mean ring.

4.5. Interatomic correlation of indole ring

In order to investigate the possibility of 'concerted' atomic movement in the indole ring, correlations between bond lengths and between bond angles were calculated using data from 35 structures and the results are given in Table 5. There are four bond-length pairs and 20 bond-angle pairs exhibiting correlation coefficients larger than r(33, 0.001) = 0.532. Concerning the bond length, the C4—C5 bond showed r = 0.718 with respect to the C3-C9 bond. Since the sign of this correlation coefficient is positive, it may indicate easy stretching of the C4-C9bond in the C3-C9-C4-C5 bond sequence. On the other hand, the highest correlation coefficient (r = -0.839)

 Table 5. Correlation coefficient matrix between bond angles (upper triangle) and lengths (lower triangle),

 calculated using a data set of 35 structures

	1-2-3	2-3-9	3-9-8	4-9-8	5-4-9	4-5-6	5-6-7	6-7-8	7-8-9	1-8-9	1-8-7	3-9-4	
	-0.658	0.060	0.436	-0.766	0.715	-0.295	-0.151	0.418	0.109	-0.839	0.655	0.450	C2-N1-C8
		-0.607	-0.053	0.365	-0.421	0.195	0.060	-0.210	0.002	0.322	-0.285	-0.379	N1-C2-C3
			-0.680	0.200	0.082	-0.219	0.303	-0.064	-0.356	0.300	0.030	0.449	C2-C3-C9
N1-C8	0.227			-0.583	0.161	0.242	-0.502	0.180	0.575	-0.771	0.202	-0.321	C3-C9-C8
C2-C3	0.113	-0.093			-0.669	0.112	0.189	-0.239	-0.396	0.798	-0.371	-0.581	C4-C9-C8
C3-C9	0.493	0.448	-0.297			-0.706	0.256	0.361	-0.254	-0.513	0.695	0.617	C5-C4-C9
C4-C9	0.181	0.353	-0.241	0.154			-0.763	-0.021	0.398	0.036	-0.398	-0.372	C4-C5-C6
C4C5	0.323	0.369	-0.322	0.718	0.264			-0.531	-0.240	0.335	-0.080	0.278	C5-C6-C7
C5—C6	0.183	0.213	0.345	0.072	0.019	-0.058			-0.479	-0.332	0.718	0.097	C6-C7-C8
C6—C7	0.514	0.367	-0.125	0.235	0.584	0.216	-0.017			-0.374	-0.543	-0.108	C7—C8—C9
C7—C8	0.259	0.284	-0.283	0.579	0.283	0.572	-0.151	0.184			-0.574	-0.148	N1-C8-C9
C8C9	0.284	-0.260	0.260	-0.020	-0.333	-0.131	0.112	-0.047	-0.109			0.216	N1-C8-C7
	1-2	1-8	2-3	3-9	4-9	4-5	5-6	6-7	7-8				

observed between C2-N1-C8 was and N1-C8-C9 bond angles. Since the sign of this correlation coefficient is negative, either N1 or C8 in the bond sequence of C2-N1-C8-C9 could be thought to move. Judging from the high correlation coefficient (r = -0.771)observed between C3-C9-C8 and C9-C8-N1 bond angles. C8 could be concluded to be the most mobile atom in the indole ring. Similarly, both negative correlation coefficients of r = -0.706 and -0.763 between the neighboring angles in the C9-C4-C5-C6-C7 bond sequence indicate the easy movement of C5. A high positive correlation coefficient (r = 0.798)between C4-C9-C8 and C9-C8-N1 angles, in contrast, can be interpreted as the easy stretching of the central C8-C9 bond in the N1-C8-C9-C4 bond sequence.

5. Conclusion

The present study demonstrates the applicability of multivariate analysis to averaging of the planar ring structure in aromatic systems, using molecular dimensional data of tryptophan metabolites, and is summarized as follows.

(1) The D^2 statistic procedure, well tested by trial and error, is suitable for averaging the molecular fragments, when the number of observations available is not sufficient for the deviation of meaningful conclusions.

(2) The mean advantage of this approach is its wide range of applications to molecular fragments, particularly to aromatic rings. (3) The D^2 statistic is valuable for detecting the outliers and expressing the deformation of each molecular structure observed from the mean ring, quantitatively.

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